

J-014-1041007

Seat No. _____

B. Pharm. / M. P. M. (Sem. I) (CBCS) Examination June / July - 2019

Pharmaceutical Inorganic Chemistry: BP-104

Faculty Code: 014 Subject Code: 1041007

Time: 3 Hours] [Total Marks: 75

1 Answer the following questions:

 $10 \times 2 = 20$

- (a) Define: Radio Isotopes.
- (b) Describe Cathartics with an example.
- (c) What are Official compendia?
- (d) Clarify the term Antacid with one example.
- (e) Differentiate between Poison and Antidote.
- (f) Explain: Dentifrices and Desensitizing agents.
- (g) Describe the definition of astringent with an example.
- (h) Discuss the composition of ORS.
- (i) Write the importance of limit test.
- (j) Explain the term Buffer capacity.
- 2 Answer any two out of the following:

 $2 \times 10 = 20$

- (a) Write the assay principle and medical uses of following compounds:
 - (1) Sodium thiosulphate
 - (2) Ammonium chloride.
- (b) What are impurities? Describe the different types of impurities. Discuss sources of impurities in detail.
- (c) Explain the role of intra and extra cellular ions in maintaining physiological balance.Write a note on - Electrolyte combination theory.

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- 3 Answer any seven out of the following: $7 \times 5 = 35$
 - (a) Define limit test. Write a detailed note on limit test of arsenic.
 - (b) Discuss mechanism of action of antidote poisoning. Write a note on cyanide poisoning and its treatment.
 - (c) Explain Bronsted acid base theory. What are conjugated pairs of acid and base?
 - (d) Define half life. Explain the properties of α, β, γ radiations.
 - (e) Write a note on GM counter.
 - (f) Discuss the role of fluoride in the treatment of Dental caries.
 - (g) Write the ideal properties of Antacids. Explain assay principle and preparation of Sodium Bicarbonate.
 - (h) Describe the antimicrobial agents and classify them with suitable examples. Discuss the various mechanisms of action on them.
 - (i) Write assay principle of following:
 - (1) Chlorinated lime
 - (2) Calcium gluconate.

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